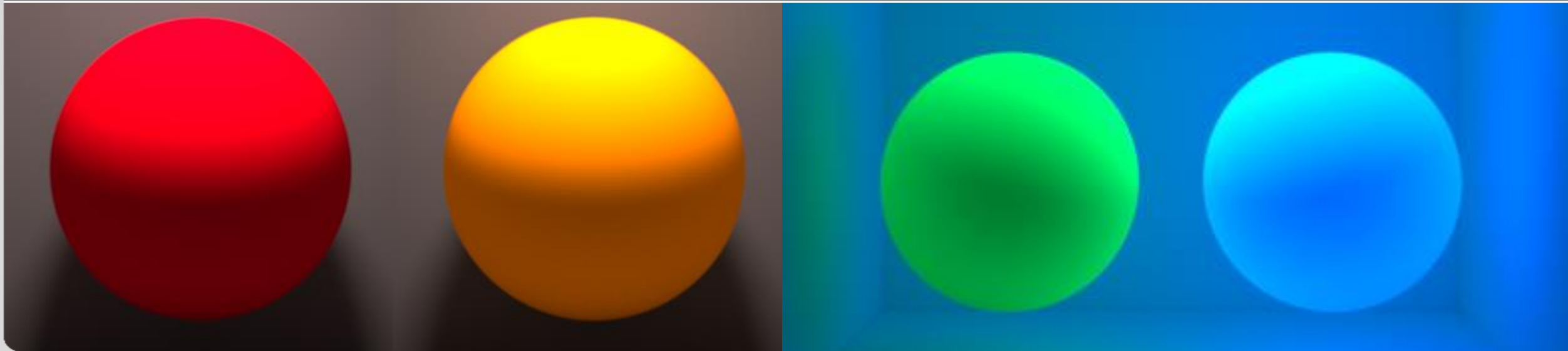
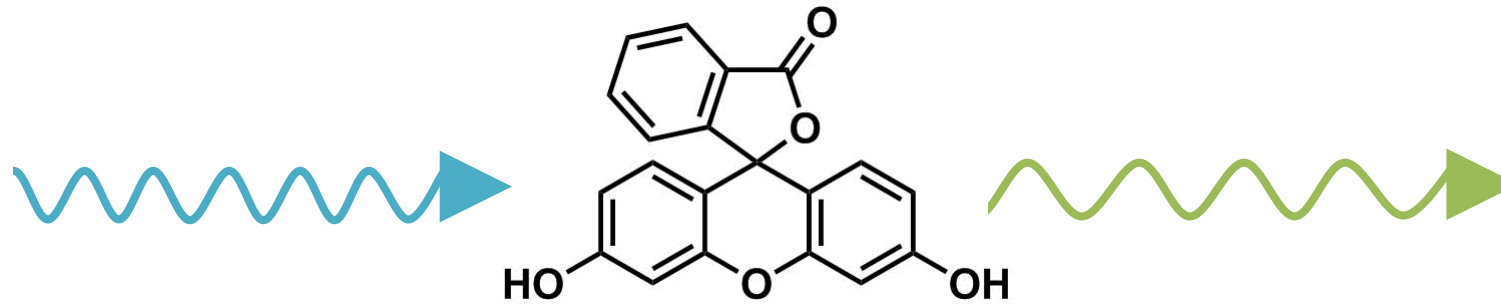


# A Simple Diffuse Fluorescent BBRRDF Model

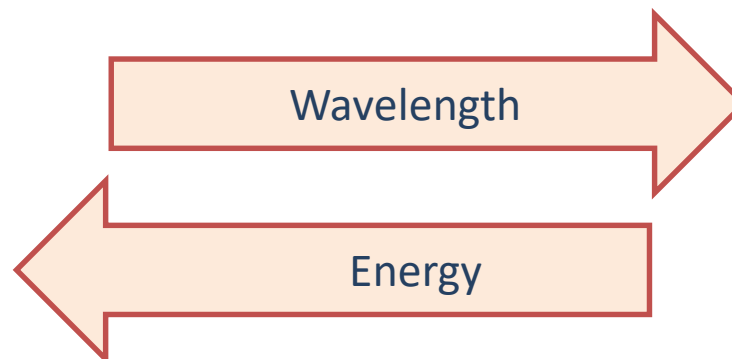
Alisa Jung, Johannes Hanika, Steve Marschner, Carsten Dachsbacher



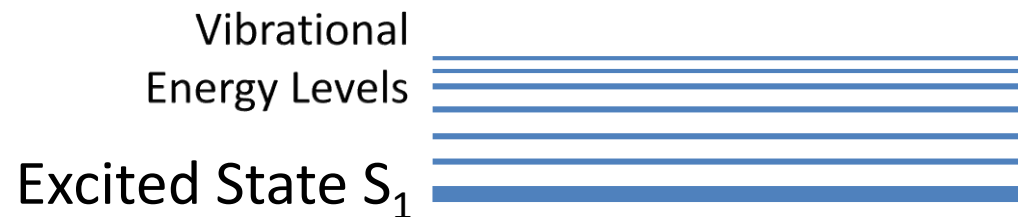
# What is fluorescence?



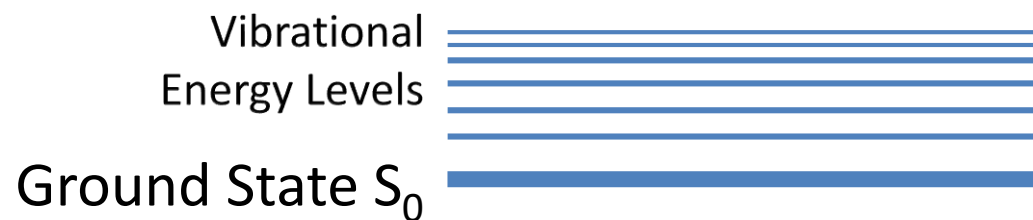
Visible Light



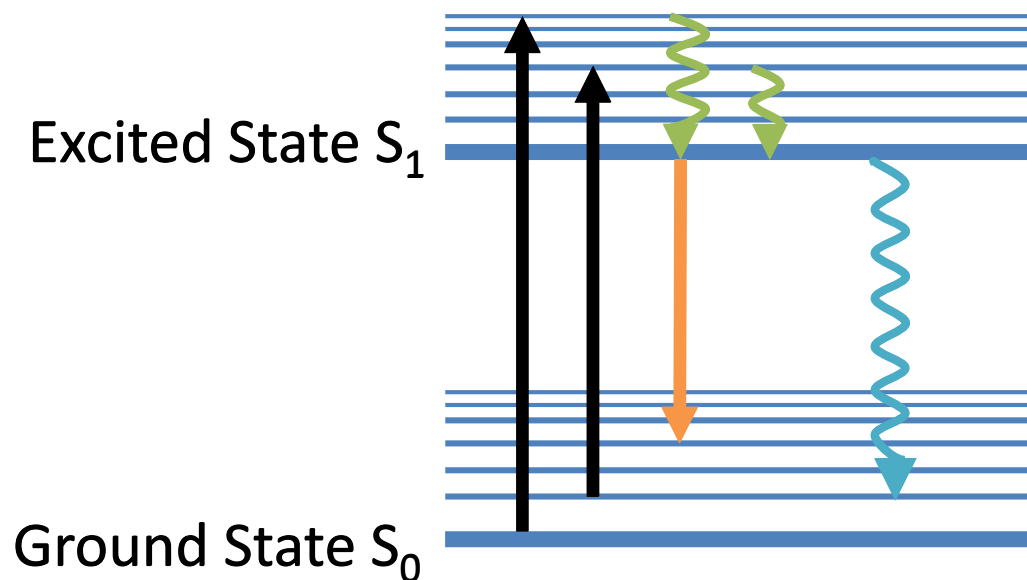
# Jablonski Energy Diagram



▶ Energy States & Levels of Molecules

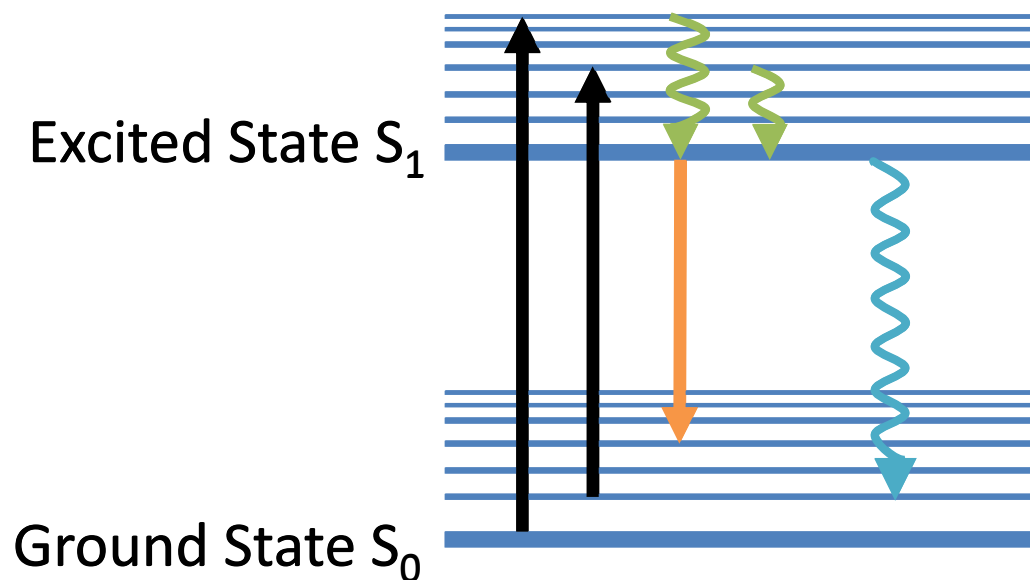


# Jablonski Energy Diagram: Fluorescence



- ▶ Absorption  $10^{-15}$  s
- ▶ Relax to  $S_1$   $10^{-12}$  s
- ▶ Fluorescence  $10^{-8}$  s
- ▶ Non-radiative Relaxation
- ▶ Diffuse
- ▶ Instantaneous
- ▶ Emitted wavelength independent of absorbed wavelength, usually longer
- ▶ Not all absorbed photons get emitted

# The Quantum Yield $\Phi$

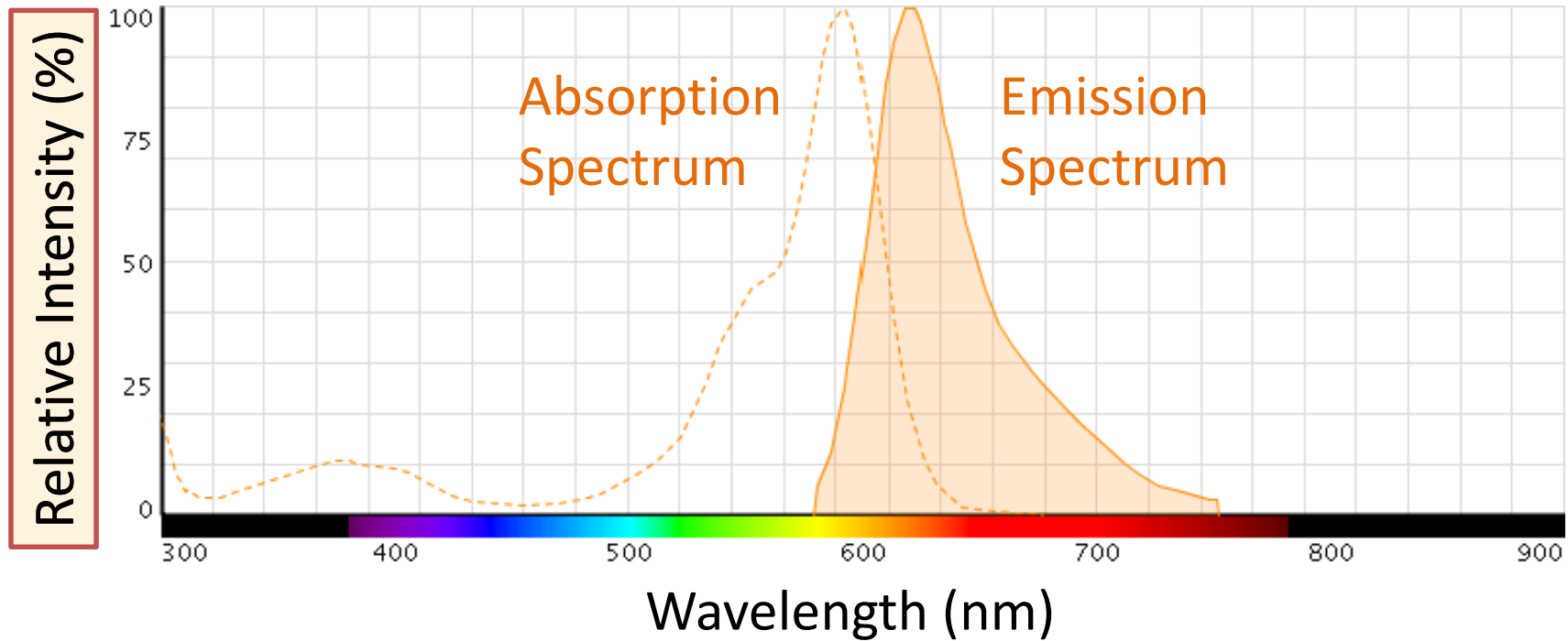


- ▶ Not all absorbed photons get emitted:

$$\Phi = \frac{\text{emitted Photons}}{\text{absorbed Photons}}$$

- ▶ Wavelength independent

# Absorption & Emission Spectrum



# The BBRRDF

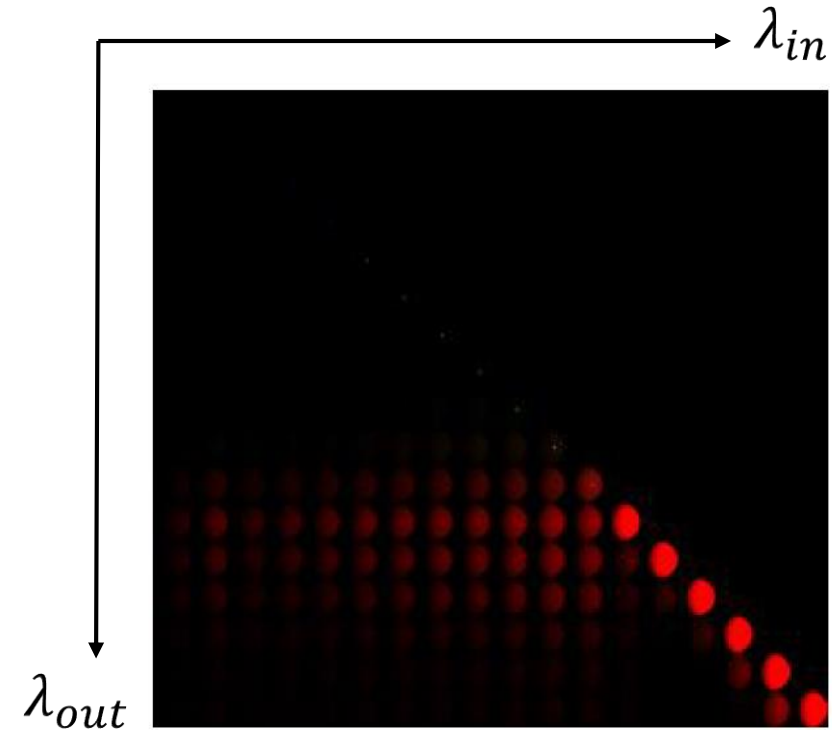
- ▶ Bispectral Bidirectional Reflection and Reradiation Distribution Function  
(Hullin et al. 2010)

$$f_r(\omega_{in}, \lambda_{in}, x, \lambda_{out}, \omega_{out}) = \frac{d^2 L_r(x, \omega_{out}, \lambda_{out})}{L_i(x, \omega_{in}, \lambda_{in}) \cos \theta_{in} d\omega_{in} d\lambda_{in}}$$

Wavelengths

# The BBRRDF: Previous Work

- ▶ Reradiation Matrix  
(*Glassner 1995, Wilkie 2001&2005, Hullin 2010*)
- ▶ Discrete description of energy shifts
- ▶ Diagonal: non-fluorescent reflectance



*Fluorescent red paint Hullin et al. 2010*

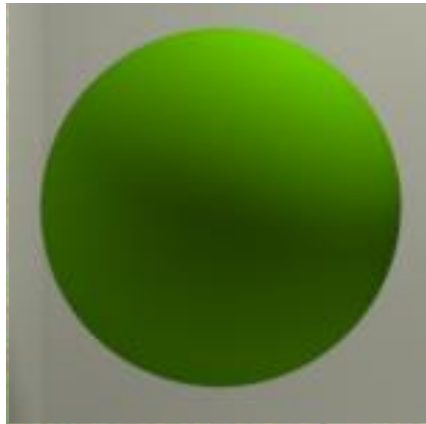


# Our BBRRDF

- ▶ Fluorescence is diffuse
  - ▶ We use a diffuse BBRRDF
- ▶ Fluorescent and non-fluorescent component
  - ▶ Light interacts either with fluorescent nor non-fluorescent molecule



Non-fluorescent



Fluorescent



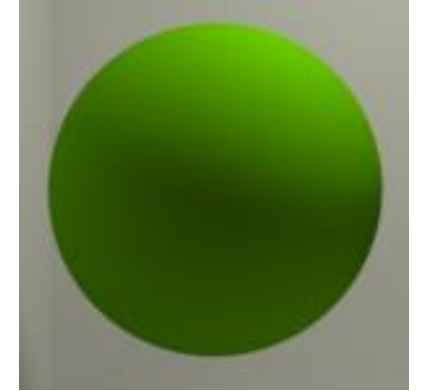
Full BBRRDF

## Our BBRRDF: Parameters

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)

# Our BBRRDF: Fluorescence

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
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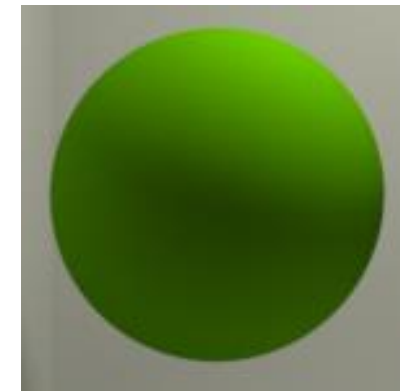


$$f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) = \boxed{c \cdot a(\lambda_{in})} \cdot Q \cdot e(\lambda_{out}) \cdot \pi^{-1}$$

Fraction of absorbed energy

# Our BBRRDF: Fluorescence

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)

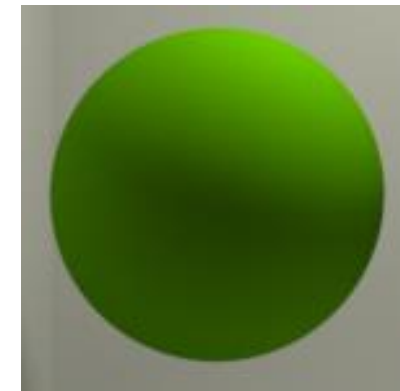


$$f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) = c \cdot a(\lambda_{in}) \cdot Q \cdot e(\lambda_{out}) \cdot \pi^{-1}$$

Fraction of emitted energy

# Our BBRRDF: Fluorescence

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)

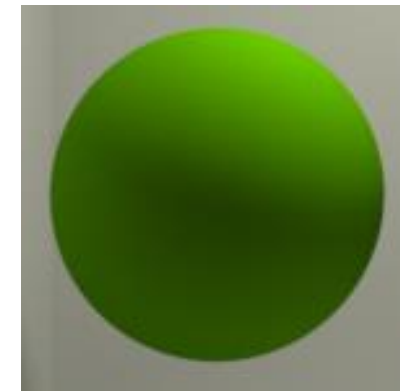


$$f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) = c \cdot a(\lambda_{in}) \cdot Q \cdot e(\lambda_{out}) \cdot \pi^{-1}$$

Fraction of emitted energy at  $\lambda_{out}$

# Our BBRRDF: Fluorescence

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)



$$f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) = c \cdot a(\lambda_{in}) \cdot Q \cdot e(\lambda_{out}) \cdot \pi^{-1}$$

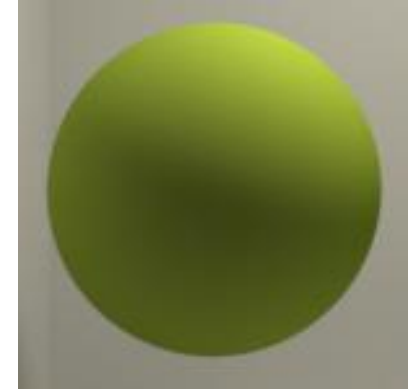
Perfectly diffuse lambert BRDF

# Our BBRRDF: Non-fluorescent Reflectance

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)

$$f(\omega_{in}, \lambda, \omega_{out}) = \boxed{(1 - c \cdot a(\lambda))} \cdot r(\lambda) \cdot \pi^{-1}$$

remaining energy



# Our BBRRDF: Non-fluorescent Reflectance

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)

$$f(\omega_{in}, \lambda, \omega_{out}) = (1 - c \cdot a(\lambda)) \cdot r(\lambda) \cdot \pi^{-1}$$

reflected energy





# Our BBRRDF

- ▶  $a(\lambda)$  absorption spectrum (scaled to 1)
- ▶  $c$  concentration parameter
- ▶  $Q$  energy quantum yield parameter
- ▶  $e(\lambda)$  emission spectrum (normed to 1)
- ▶  $r(\lambda)$  reflectance spectrum (less than 1)



$$f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) = [\delta_{\lambda_{in}, \lambda_{out}} \cdot (1 - c \cdot a(\lambda_{in})) \cdot r(\lambda_{in}) + c \cdot a(\lambda_{in}) \cdot Q \cdot e(\lambda_{out})] \cdot \pi^{-1}$$

# Our BBRRDF: Wavelength Sampling

► Delta component → 2 Steps:

► Sample if light interacts with fluorescent particle

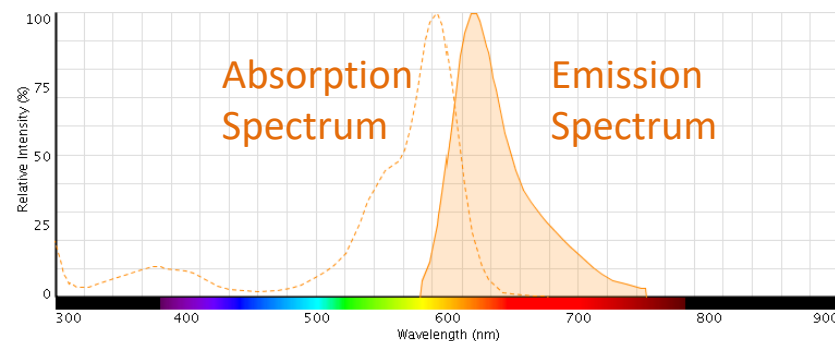
$$P(\text{fluorescence}) = \frac{\text{fluorescently reflected energy}}{\text{total reflected energy}}$$

► If so, sample new wavelength

Camera path:  $p(\lambda_{in}) \propto a$

Light path:  $p(\lambda_{out}) \propto e$

► Different for camera and light paths!



# Energy Conservation vs. Photon Conservation

▶ Energy Conserving BRDF:

$$\forall \omega_{in} \in \Omega: \int_{\Omega} f(\omega_{in}, \omega_{out}) d\omega_{out}^{\perp} \leq 1$$

# Energy Conservation vs. Photon Conservation

▶ Energy Conserving BBRRDF:

$$\forall \omega_{in} \in \Omega, \lambda_{in} \in \Lambda: \int_{\Omega \times \Lambda} f(\omega_{in}, \lambda_{in}, \lambda_{out}, \omega_{out}) d(\omega_{out}^{\perp}, \lambda_{out}) \leq 1$$

▶ Our BBRRDF is energy conserving if

▶  $a(\lambda), r(\lambda), Q, c \in [0,1]$

▶  $\int e(\lambda) d\lambda = 1$

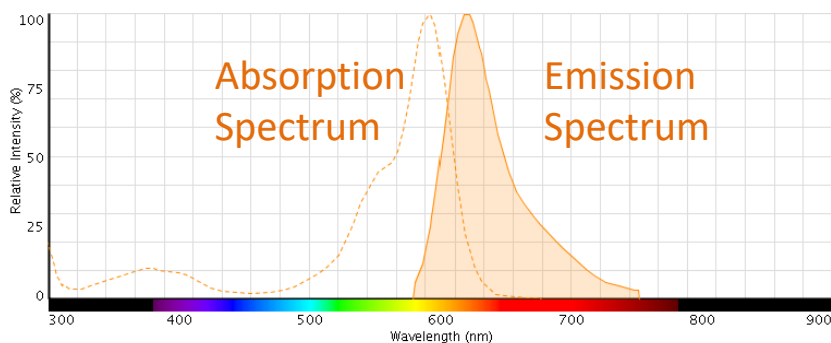
▶ But it is not yet photon conserving!

# Energy Conservation vs. Photon Conservation

Example:

▶  $Q = 1$

▶  $c = 1$



▶ Consider  $\lambda_{in}$  where  $a(\lambda_{in}) = 1$

Energy of a Photon:

$$E = \frac{hc}{\lambda}$$

